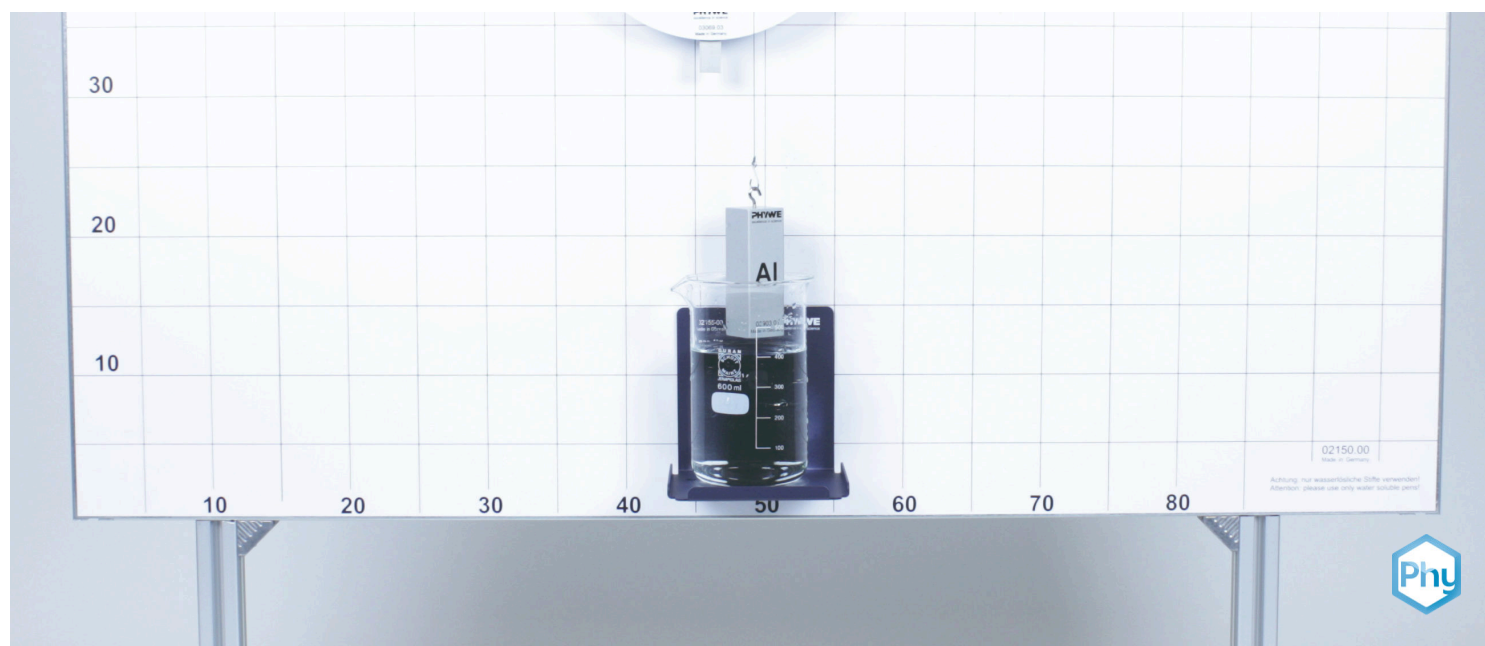


# Density determination by measuring the buoyancy



P1297300

Physics

Mechanics

Mechanics of liquids &amp; gases



Difficulty level

medium



Group size

-



Preparation time

10 minutes



Execution time

10 minutes

This content can also be found online at:

<http://localhost:1337/c/66d8307af284a400027dc679>

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## General information



## Application (1/2)

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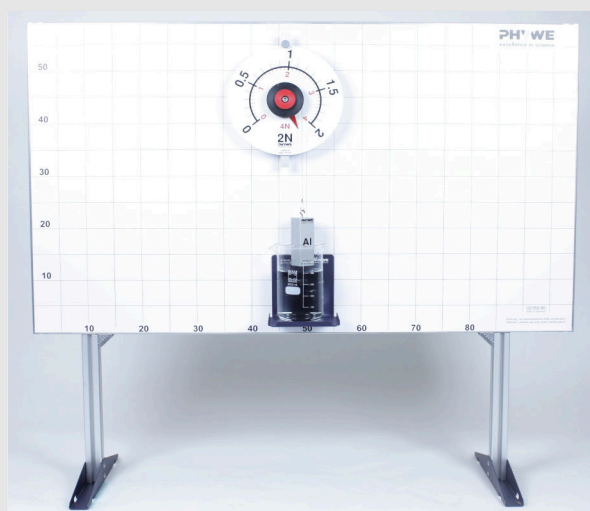


Fig. 1: Experimental setup

According to Archimedes' principle, the density of solid and liquid substances can be determined by their buoyancy.

This principle is used, for example, in a hydrometer. The hydrometer is a measuring device for determining the density or specific weight of liquids. This is based on Archimedes' principle: a body is immersed in a liquid until the weight of the displaced liquid corresponds to the weight of the immersed body.

## Application (2/2)

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This has two consequences:

1. The lower the density of the liquid, the further a body of the same weight is immersed in it. (Hydrometer)
2. If a body is supposed to sink to a certain point in liquids of different densities or different specific weights, its weight must be artificially increased to the same extent as the density increases. (Density meter)

## Other information (1/2)

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### Prior knowledge



The students should have prior knowledge of "Archimedes' principle" and buoyancy.

### Principle



The aim is to show how the density of solid and liquid substances can be determined by measuring buoyancy according to Archimedes' principle.

## Other information (2/2)

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### Learning objective



The students are supposed to learn how to determine density by measuring buoyancy and get a deeper understanding of Archimedes' principle.

### Tasks



The students should determine the density of solid and liquid substances by measuring the buoyancy. In addition, they can determine the density according to the obtained density ( $\rho$ ) to determine the substances.

## Safety instructions

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The general safety instructions for experimentation in science lessons apply.

## Theory (1/2)

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The density of a homogeneous substance is defined by the ratio of its mass  $m$  and its volume  $V$ :

$$\rho = m/V$$

Before the density ( $\rho$ ) is determined, the mass ( $m$ ) and the volume ( $V$ ) should be known.

## Theory (1/2)

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### 1st experiment: Determining the density of solids

The mass ( $m$ ) can be known by the buoyancy force. When a body floats, two forces act on it: the weight force ( $F_G$ ) which pulls it downwards, and the the upward buoyancy force ( $F_A$ ). This means that the body is only immersed in the liquid until the two forces equalise.

Therefore applies:

$$F_A = F_G \rightarrow \rho_{F_l} \cdot A \cdot h \cdot g = m \cdot g$$

The volume can be determined by the mass of the displaced water.

## Theory (2/2)

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### 2nd experiment: Determining the density of liquid substances

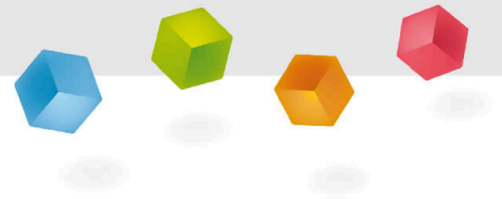
The volume (  $V$  ) is known. And the mass (  $m$  ) can be determined by the buoyancy force because, according to Archimedes' principle, a body experiences an upward force equal to the weight of the fluid it displaces when it is completely or partially immersed in a fluid.

## Equipment

Position	Equipment	Item no.	Quantity
1	<a href="#">PHYWE Demo Physics board with stand</a>	02150-00	1
2	<a href="#">Torsion dynamometer</a>	03069-03	1
3	<a href="#">Support plate on fixing magnet</a>	02155-00	1
4	<a href="#">Sinkers, aluminium</a>	03903-01	1
5	<a href="#">Beakers, Boro, high form, various sizes (600 ml)</a>	46029-00	1
6	<a href="#">Glycerine, 250 ml</a>	30084-25	2
7	<a href="#">Screw clamp</a>	02014-01	2

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## Setup and procedure



### Setup and procedure (1/2)

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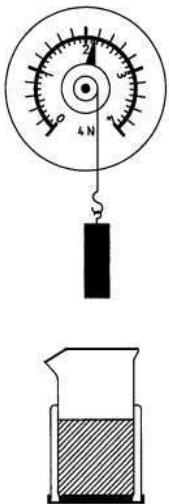


Fig. 2

#### 1st experiment

- Place the torsion dynamometer on the upper edge of the demo board.
- Suspend the sinker and measure and record its weight in air ( $F_{G,L}$ ).
- Place the support plate underneath the sinker at the bottom edge of the board and place the beaker with approx. 400 ml of water on it (Fig. 2).
- Lower the dynamometer with the sinker until the latter is completely immersed in the water.
- Measure the force  $F_{G,W}$  with which the sinker now pulls on the dynamometer and note  $F_{G,W}$ .



## Setup and procedure (2/2)

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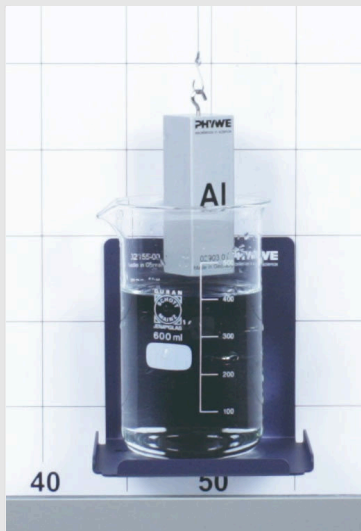


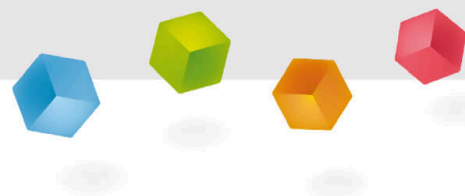
Fig. 3

### 2nd experiment

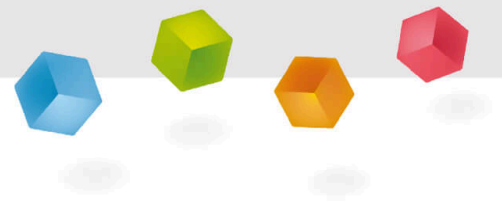
- Leave the setup as in experiment 1 but now replace the water in the beaker with ethanol.
- Lower the torsion dynamometer with the sinker attached until it is completely immersed in the alcohol.
- Measure the force  $F_{G,Sp}$  with which the sinker pulls on the dynamometer and note  $F_{G,Sp}$ .
- Add glycerol to the beaker instead of ethanol and determine the force  $F_{G,Gl}$  in the same way.

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## Evaluation



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# Evaluation

## Observation

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### 1st experiment

$$F_{G,L} = 2,15N$$

$$F_{G,W} = 1,35N$$

### 2nd experiment

$$F_{G,sp} = 1,47N$$

$$F_{G,Gl} = 1,19N$$

$$F_{G,L} = 2,15N$$

$$V = 82cm^3$$

## Evaluation (1/3)

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### 1st attempt

The buoyancy force acting on the sinker is

$$F_a = F_{G,L} - G_{G,W} = 0,8N$$

$F_A$  is equal to the weight of the water displaced by the immersed body according to Archimedes' law. The mass of the displaced water is therefore  $m_W = 82g$  ( $1N = 102g$ ) and its volume is  $V_W = 82cm^3$ . The sinker therefore has the volume  $V = V_W = 82cm^3$  and because of  $F_{G,L} = 2,15N$  the mass is  $m = 219g$

For the required density of aluminium this results in:

$$\rho = m/V = 219g/82cm^3 = 2,7g/cm^3$$

This corresponds to the table value.

## Evaluation (2/3)

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Liquid	$V/cm^3$	$F_A/N$	$m/g$	$\frac{\rho}{g/cm^3}$
Water	82	0,80	82	1,00
Ethanol	82	0,68	69	0,84
Glycerol	82	0,96	98	1,20

Tab. 1

The value measured in experiment 1 for  $F_{G,L}$  as well as the one calculated for the volume  $V$  ( $V = V_W = V_{Sp} = V_{Gl}$ ) are accepted as results. Now the buoyancy force  $F_A$  for ethanol and glycerol and the from this resulting masses of the displaced liquids are calculated and finally the densities.

It is advisable to summarise all the results from experiments 1 and 2 in a table (see Table 1):

$$m = F_G/g,$$

$$\rho = m/V$$

## Evaluation (3/3)

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Within the scope of the measurement accuracy the results agree well with the table values for the densities :

$$\rho_{Sp} = 0,85g/cm^3,$$

$$\rho_{Gl} = 1,20g/cm^3.$$

## Notes (1/2)

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### 1st experiment

To simplify matters and within the limits of measurement accuracy, it can be assumed that 1 N corresponds to the weight force acting on a body with the mass  $m = 100g$  and the evaluation facilitated:

$$F_A = 0,80N \rightarrow m_W = 80g \rightarrow V_W = V = 80cm^3;$$

$$F_{G,L} = 2,15N \rightarrow m = 215;$$

$$\rho = m/V = 215g/80cm^3 = 2,7g/cm^3$$

If the students do not know that the sinker is made of aluminium, in addition to the determination of  $\rho$  the experiment can also be used to determine the possible material of the sinker. In this case the symbol on the sinker should be taped over before the experiment.