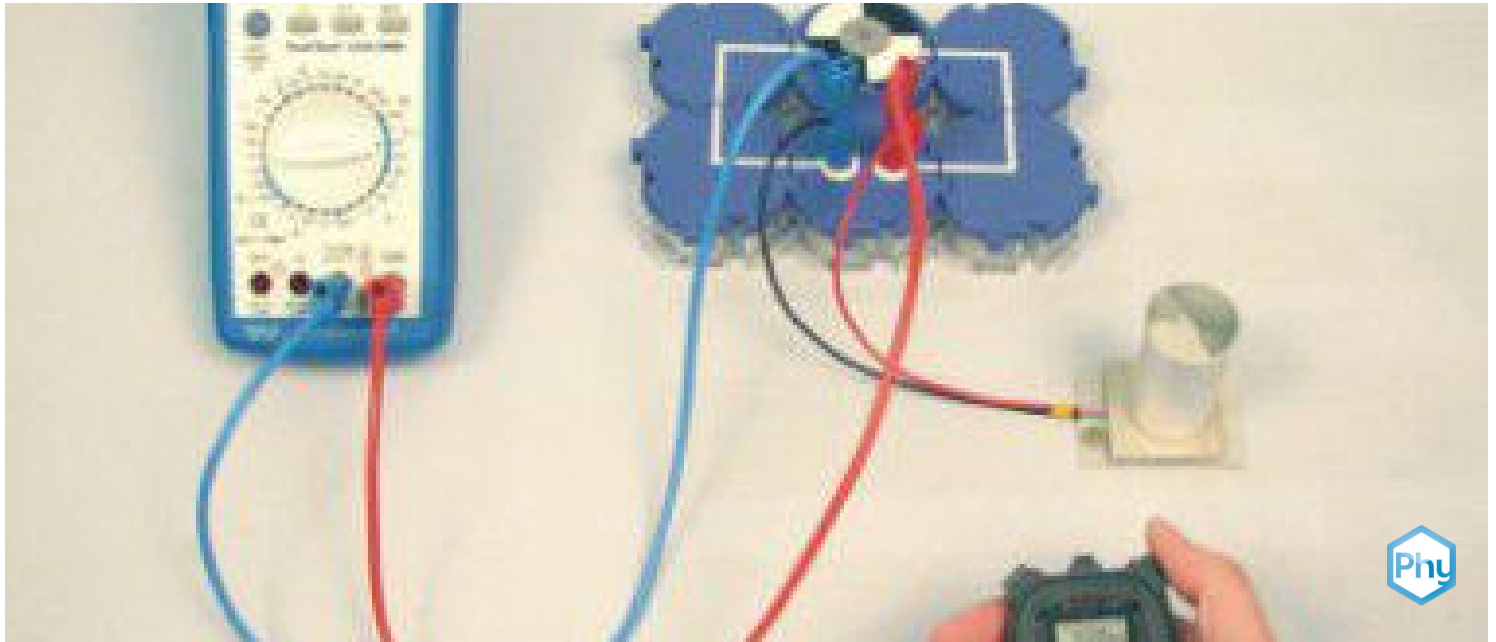


# Generation of electrical energy using a thermogenerator (thermoelectric power)



Physics

Energy

Renewable energies: Earth



Difficulty level

easy



Group size

1



Preparation time

10 minutes



Execution time

10 minutes

This content can also be found online at:



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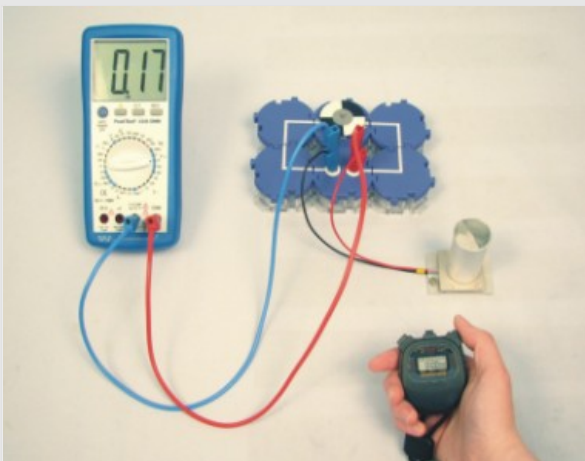
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## Teacher information



## Application

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Experimental setup

The Peltier element (the thermogenerator) consists of many thermocouples. These are connected electrically in series and thermally in parallel so that their thermoelectric voltages add up.

This experiment shows that a thermogenerator can generate electrical energy from thermal energy. The thermoelectric voltage of a thermogenerator depends on the temperature difference across the element.

A large storage tank (aluminium block) can keep the temperature difference stable (and thus higher) over a longer period of time and therefore has advantages in energy production.

## Other teacher information (1/3)

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### Prior knowledge



Students should be familiar with the basic concepts of thermodynamics.

### Principle



In this experiment, a Peltier element is put into operation and investigated to see how it produces an electric current and what properties this current has.

## Other teacher information (2/3)

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### Learning objective



### Tasks



Students learn how to generate electricity with a thermogenerator.

- The thermogenerator consists of a so-called "Peltier element", which is mounted between two thin aluminium plates.
- Observe what happens when the aluminium plates are heated or cooled.
- Try to run a small motor with the thermogenerator.
- In addition, there is a large aluminium block. Investigate what significance it has.

## Other teacher information (3/3)

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### Notes on set-up and procedure

The experiment consists of several small experiments. Make sure to leave enough time between these experiments for the two sides of the thermogenerator to cool down to room temperature again and thus for the thermoelectric voltage to approach zero.

For each group of students, provide small pieces of ice that fit into the blank aluminium cup.

The results depend on the ambient temperature, so the measured values may deviate from the sample solution. However, the general course of the thermoelectric voltage remains the same.

## Safety instructions

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The general instructions for safe experimentation in science lessons apply to this experiment.

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## Student information

### Motivation

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A campfire

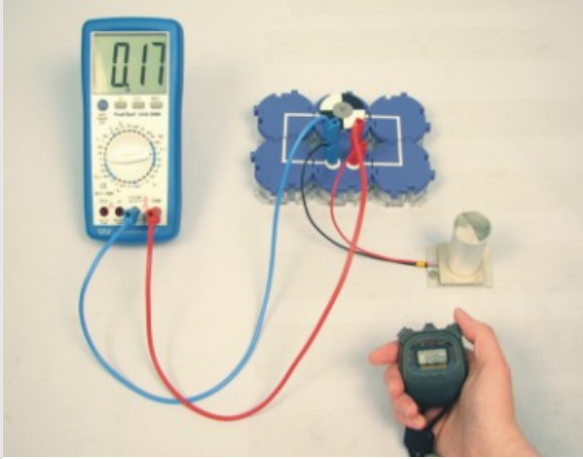
Heat is one of the most intuitive and ubiquitous forms of energy and is also a by-product of almost all physical processes.

Using thermogenerators, it is possible to convert this common form of energy into electricity and meet the ever-growing demand for electronics in everyday life.

A typical example of a thermogenerator is the Peltier element, which uses temperature differences to generate a current flow.

## Tasks

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The experimental setup

The thermogenerator consists of a so-called "Peltier element", which is mounted between two thin aluminium plates.

Observe what happens when the aluminium plates are heated or cooled.

Try to run a small motor with the thermogenerator.

In addition, there is a large aluminium block. Investigate what significance it has.

## Equipment

Position	Material	Item No.	Quantity
1	Thermal generator for student experiments	05770-00	1
2	Motor with indicating disc, SB	05660-00	1
3	Angled connector module, SB	05601-02	4
4	Interrupted connector module with sockets, SB	05601-04	1
5	Beaker, aluminum, polished	05903-00	1
6	Beaker, 100 ml, plastic (PP)	36011-01	1
7	Lab thermometer, -10...+110 °C	38056-00	1
8	Digital stopwatch, 24 h, 1/100 s and 1 s	24025-00	1
9	Connecting cord, 32 A, 500 mm, red	07361-01	1
10	Connecting cord, 32 A, 500 mm, blue	07361-04	1
11	PHYWE Digital multimeter, 600V AC/DC, 10A AC/DC, 20 MΩ, 200 μF, 20 kHz, -20°C...760°C	07122-00	1

## Set-up

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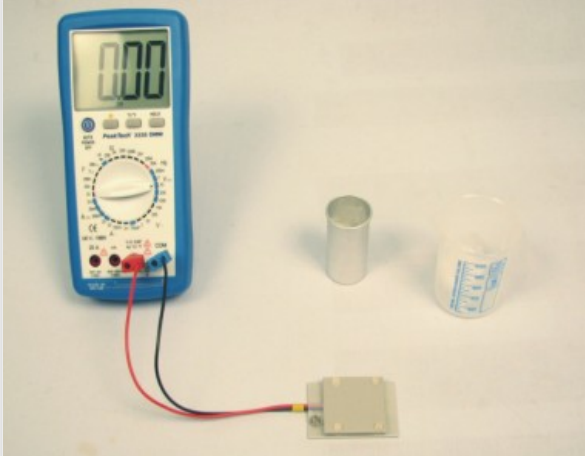


Figure 1

1. You need small pieces of ice for the experiment. Keep them in the 100 ml beaker.
2. Fill the blank cup full with small ice cubes and add some melted water or cold water (the cup should be filled no more than halfway with water).
3. Place the thermogenerator on the table so that the larger plate is at the bottom and connect the element to the voltage input of the meter (Fig. 1). Select the DC voltage measuring range 2V-.

## Procedure (1/3)

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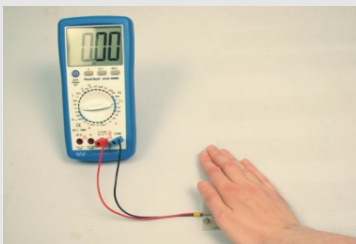


Figure 2



Figure 3

1. Only touch the upper plate of the thermogenerator with your hand (Fig. 2). Note in the report whether the indicated voltage is positive or negative.
2. Wait until the voltage is approximately between -10 mV and +10 mV again.  
Place the blank beaker with ice on the thermogenerator (Fig. 3). Observe and note whether the voltage displayed is positive or negative and take the beaker back off the thermogenerator.
3. Turn the thermogenerator over so that the smaller plate is on the table. Wait until the voltage is again approximately between -10 mV and +10 mV. Now repeat the experimental steps just described.



## Procedure (2/3)

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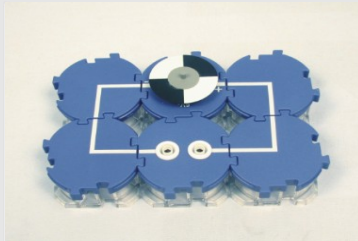


Figure 4

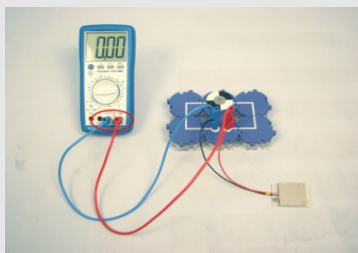


Figure 5

**4.** Build the circuit for the motor according to Figure 4.

**5.** Place the thermogenerator on the table so that the larger plate is at the bottom. Connect the motor and the thermogenerator according to Figure 5.

Swap the voltage inputs on the meter so that positive values are displayed.

**6.** Stir the ice water in the blank beaker until the thermometer shows about 0 °C to 1 °C (fig. 6). Add a few pieces of ice if necessary.



Figure 6

## Procedure (3/3)

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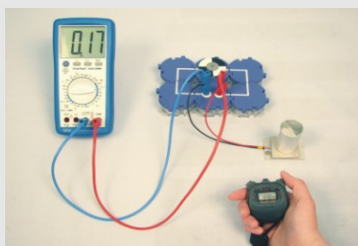


Figure 7

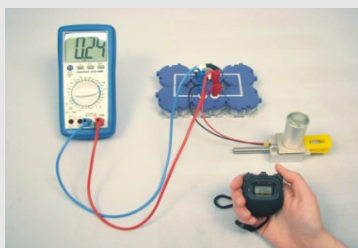


Figure 8

**7.** Place the cup on the thermogenerator and start the stopwatch (Fig. 7). Observe the meter and the motor, possibly prodding the motor. Note the voltage and your observation at 15 s intervals.

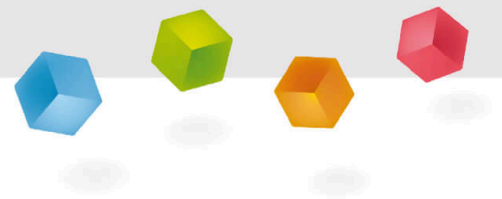
**8.** Take the beaker from the thermogenerator. Warm the thermogenerator back up to room temperature with your hands and wait until the voltage is approximately between -10 mV and +10 mV again. Place the thermogenerator on the aluminium block and clamp it.

Observe the voltage and possibly wait again for it to return to zero.

Stir the ice water in the beaker again and place the beaker on the thermogenerator with aluminium block (Fig. 8).

**9.** Repeat the experiment and note down the results.

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# Report

## Task 1

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### Drag the words into the correct gaps

The  is used to create a temperature difference between two  via the Peltier effect. To do this, an  must be passed through the semiconductor plates. However, it is possible to reverse this effect by first creating the  and then generating the current flow from it. This phenomenon is called the .

Peltier element

Seebeck effect

electrical current

temperature difference

semiconductor plates

☒ Check

## Task 2

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What does the material- and temperature-dependent Seebeck coefficient describe?

$$S_m(T) = \frac{\boxed{\phantom{000000}}}{\boxed{\phantom{000000}}}$$

$t$	$\eta$	$v$
$U$	$A$	$\rho$
$I$	$\Delta T$	$m$

$t$  = Time,  $\eta$  = efficiency,  $v$  = speed,  $U$  = Voltage,  $A$  = contact surface,  $\rho$  = Air density,  $I$  = current intensity,  $\Delta T$  = temperature difference,  $m$  = mass

✓ Check

## Task 3

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Which two of these equations describe the voltage  $U$  in a Peltier element?

☐  $U = \frac{S_1 - S_2}{\Delta T}$

☐  $U = \int_{\Delta T} (S_1(T) - S_2(T)) dT$

☐  $U = (S_1 - S_2) \cdot \Delta T$  at constant outside temperature and small  $\Delta T$

☐  $U = \frac{S_1}{S_2}$

✓ Check

Slide	Score / Total
Slide 16: Seebeck effect phenomenon	0/5
Slide 17: Seebeck coefficient	0/2
Slide 18: Peltier element	0/2

Total  0/9



Solutions



Repeat